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**Refined atomic coordinates for the sillimanite structure.** By SLAVOMIL ĎUROVIČ, *Refractories Research Institute, Bratislava, Czechoslovakia*, and ŠTEFÁNIA DÁVIDOVÁ, *Department of Mineralogy and Crystallography, Faculty of Natural Sciences, Komenský University, Bratislava, Czechoslovakia*

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In the course of a comparison of the structure of mullite (Ďurovič, 1962a) with that of sillimanite (Ďurovič, 1962b), it was necessary to re-examine the sillimanite structure of Hey & Taylor (1930). One of the main results of the mullite structure analysis was the establishment of statistical occupation of the Al, Si\* atomic positions (Fig. 1(a)). This statistical occupation was indicated by low peaks in the Fourier projection  $\rho(x, y)$ , and confirmed during the refinement. The same result was obtained independently by Sadanaga, Tokonami & Takéuchi (1962). It was therefore desirable to determine whether the sillimanite structure showed a similar effect, especially as the experimental methods used by Hey & Taylor were insufficiently sensitive for this purpose.

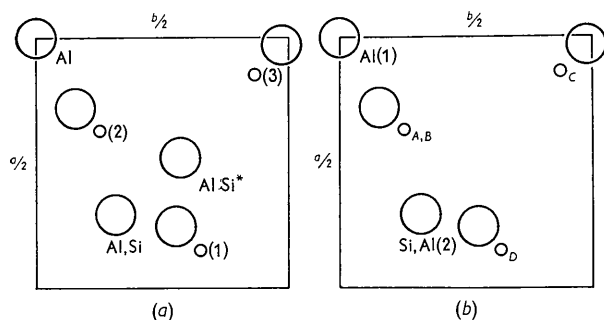


Fig. 1. Atomic positions in the mullite (a) and sillimanite (b) crystal structures.

For the re-examination crystals of the sillimanite variety xenolite (40.30% SiO<sub>2</sub>, 59.60% Al<sub>2</sub>O<sub>3</sub>, 0.85% Fe<sub>2</sub>O<sub>3</sub> and 0.07% TiO<sub>2</sub>; ideal composition 62.93% Al<sub>2</sub>O<sub>3</sub>) were used. The measured density was  $3.226 \pm 2$  g.cm.<sup>-3</sup>. The electron-density projection was calculated from 72 independent reflexions observed on Weissenberg photographs taken with Cu radiation. The structure was refined by repeated  $F_o - F_c$  syntheses, the final  $R$  factor being 0.071 (0.068). The syntheses showed no Al, Si\* maximum

of the type found in mullite (Fig. 1(b)). The final atomic coordinates are compared with those determined by Hey & Taylor in Table 1. The assignment of the  $z$

Table 1. Refined atomic coordinates of sillimanite and those determined originally by Hey & Taylor

Atoms	This work			Hey & Taylor		
	$x$	$y$	$z$	$x$	$y$	$z$
Al(1)	0	0	0	0	0	0
Al(2)	0.352 $\pm 0.001$	0.158 $\pm 0.001$	$\frac{1}{2}$ ( $\frac{1}{2}$ )	0.36	0.15	$\frac{1}{2}$
Si	0.352 $\pm 0.001$	0.158 $\pm 0.001$	$\frac{1}{2}$ ( $\frac{1}{2}$ )	0.36	0.15	$\frac{1}{2}$
O <sub>A</sub>	0.139 $\pm 0.002$	0.079 $\pm 0.002$	$\frac{1}{2}$	0.15	0.07	$\frac{1}{2}$
O <sub>B</sub>	0.139 $\pm 0.002$	0.079 $\pm 0.002$	$\frac{1}{2}$	0.15	0.07	$\frac{1}{2}$
O <sub>C</sub>	0.025 $\pm 0.002$	0.490 $\pm 0.002$	$\frac{1}{2}$	0.03	0.47	$\frac{1}{2}$
O <sub>D</sub>	0.374 $\pm 0.002$	0.275 $\pm 0.002$	0	0.39	0.28	0

parameters was verified by comparison of observed and calculated values of  $F(h0l)$  and  $F(0kl)$ .

No detailed analysis of the occupation of the tetrahedral positions was possible.

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**The crystal structure of SnCl<sub>2</sub>. A correction.** By J. M. VAN DEN BERG, *Laboratory of Inorganic Chemistry, University of Leiden, The Netherlands.*

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The text of the paper by van den Berg, 1961, contains an error in the interatomic distances of the Sn atom at  $z = \frac{1}{4}$  to the Cl atoms in the same plane. These distances should be 3.22, 2.66 and 3.30 Å. The distances in Fig. 1 are correct.

#### Reference

- BERG, J. M. VAN DEN (1961). *Acta Cryst.* **14**, 1002.